



12-2017

Evolution Characteristics of Perovskite Solar Cells in Air and Vacuum Environments

Xiaohui Wang

University of Kentucky, bxezwxh@126.com

Hojjatollah Sarvari

University of Kentucky, hojjat.sarvari@uky.edu

Hongmei Dang

University of Kentucky, hdang2@g.uky.edu

Zhi Chen

University of Kentucky, zhi.chen@uky.edu

Vijay P. Singh

University of Kentucky, vsingh@uky.edu

Right click to open a feedback form in a new tab to let us know how this document benefits you.

Follow this and additional works at: https://uknowledge.uky.edu/ece_facpub

 Part of the [Electrical and Computer Engineering Commons](#)

Repository Citation

Wang, Xiaohui; Sarvari, Hojjatollah; Dang, Hongmei; Chen, Zhi; and Singh, Vijay P., "Evolution Characteristics of Perovskite Solar Cells in Air and Vacuum Environments" (2017). *Electrical and Computer Engineering Faculty Publications*. 31.
https://uknowledge.uky.edu/ece_facpub/31

This Article is brought to you for free and open access by the Electrical and Computer Engineering at UKnowledge. It has been accepted for inclusion in Electrical and Computer Engineering Faculty Publications by an authorized administrator of UKnowledge. For more information, please contact UKnowledge@lsv.uky.edu.

Evolution Characteristics of Perovskite Solar Cells in Air and Vacuum Environments

Notes/Citation Information

Published in *Optik*, v. 150, p. 111-116.

© 2017 Elsevier GmbH. All rights reserved.

This manuscript version is made available under the CC-BY-NC-ND 4.0 license

<https://creativecommons.org/licenses/by-nc-nd/4.0/>.

The document available for download is the author's post-peer-review final draft of the article.

Digital Object Identifier (DOI)

<https://doi.org/10.1016/j.ijleo.2017.09.100>

Accepted Manuscript

Title: Evolution characteristics of perovskite solar cells in air and vacuum environments

Authors: Xiaohui Wang, Hojjatollah Sarvari, Hongmei Dang, Zhi Chen, Vijay Singh



PII: S0030-4026(17)31170-1
DOI: <https://doi.org/10.1016/j.ijleo.2017.09.100>
Reference: IJLEO 59708

To appear in:

Received date: 14-6-2017
Accepted date: 25-9-2017

Please cite this article as: Xiaohui Wang, Hojjatollah Sarvari, Hongmei Dang, Zhi Chen, Vijay Singh, Evolution characteristics of perovskite solar cells in air and vacuum environments, *Optik - International Journal for Light and Electron Optics* <https://doi.org/10.1016/j.ijleo.2017.09.100>

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

Evolution characteristics of perovskite solar cells in air and vacuum environments

Xiaohui Wang^{a,b*}, Hojjatollah Sarvari^b, Hongmei Dang^b, Zhi Chen^{a,b}, and Vijay Singh^b

^a School of Optoelectronic Information, University of Electronic Science and Technology of China, Chengdu, Sichuan 610054, China

^b Department of Electrical & Computer Engineering and Center for Nanoscale Science & Engineering, University of Kentucky, Lexington, Kentucky 40506, USA

Abstract:

Solid state hybrid organometal halide perovskite solar cells have become the research hotspots recently. We prepared perovskite solar cells by the simple two-step method in the ambient condition. The solar cell samples were kept in air and vacuum environments, respectively. J_{sc} shows good stability in vacuum for up to 27 days, while it drops a lot in air. There is a remarkable promotion of FF for the air-stored samples compared that of the vacuum-stored samples. Changes in V_{oc} of vacuum-stored and air-stored samples exhibited a similar behavior, both showing a slight increase. We believe the moisture in air makes perovskite decompose which leads to the decrease of J_{sc} , and the HTM material can be oxidized in air and its conductivity will be improved, which is beneficial to the FF.

Keywords:

Perovskite solar cells, Evolution, Ambient atmosphere, Vacuum

1. Introduction

The methylammonium lead halide ($\text{CH}_3\text{NH}_3\text{PbX}_3$, X = halogen; CH_3NH_3 : MA) based perovskite solar cells stand out from many new solar cells in recent years, and attract the attention of many scientific researchers¹⁻⁶. Since the proposal of being a light sensitizer for photovoltaic cells by Miyasaka *et al.* in 2009⁷, the power conversation efficiency (PCE) record exceeds 20% for the perovskite solar cells recently^{8, 9}. Compared to other solar cells, perovskite solar cells shows several appealing features, including intense light absorption, excellent ambipolar charge mobility, small binding energy, high efficiency, and low fabrication cost^{10, 11}.

The PCE of perovskite reached a good level even compared with the traditional solar cells, but the stability of perovskite is poor, especially contact with moisture in air. So besides the pursuit of high PCE, it is also important to know the evolution characteristics of the perovskite solar cells and improve its stability. There are some research on stability of perovskite solar cells with unique structure or composition. With the optimized thickness of the mesoporous TiO_2 layer, the PCE can be maintained over

* Corresponding author. Xiaohui Wang E-mail address: bxezwxh@126.com

85% of the initial value after 100 days for the perovskite solar cells stored in air condition¹⁰. By introducing an efficient pristine hole-transporting material (HTM), tetrathiafulvalene derivative without the use of tert-butylpyridine and lithium salts, the stability of the dopant-free perovskite solar cells was greatly improved two fold in air¹¹. Baohua Wang *et al.* prepared perovskite films by a modified chemical vapor transport approach, and the perovskite solar cells showed a stable characteristics for 100 days in air¹². The relative stability under the same conditions in air also can be greatly improved by using optimal mixed-halide compositions¹³.

Structure of the solar cells, dopant for HTM, preparation method, and halogen composition are found to be important for the stability of perovskite solar cells, but in all of the previous studies, effects of the atmosphere cannot be ignored. However, the role played by the atmospheric environment on the evolution of perovskite solar cells has rarely been reported. Here we fabricated perovskite solar cells with the simple two-step deposition technique in the ambient condition¹⁴⁻¹⁷. To study the role played by the atmosphere environment on the perovskite solar cells evolution, the prepared samples were stored in air and vacuum for several days, respectively. The current density (J_{sc}), open circuit voltage (V_{oc}), fill factor (FF), and PCE under different conditions were recorded, compared, and discussed.

2. Experimental section

2.1 Materials

All materials were purchased from Sigma-Aldrich, unless otherwise specified, and they were used as received. Methylammonium iodide ($\text{CH}_3\text{NH}_3\text{I}$), Co(III) TFSI Salt, and Spiro-MeOTAD were purchased from Luminescence Technology.

2.2 Fabrication of perovskite solar cells

Here, the solar cells were fabricated using the widely used two-step method. FTO-coated glass substrates were cut in the dimensions of 1×1 inch. The surround conductive layer was patterned by lithography, then etched using zinc powder and hydrochloric acid followed by diluted hydrofluoric acid, to prevent the leakage on the side. Then the FTO substrate was sequentially washed with acetone, isopropanol, Potassium hydroxide in an ultrasonic bath for 10 min respectively, rinsed with distilled water and dried by nitrogen finally. A compact thin layer of TiO_2 was then deposited by RF plasma sputtering using a TiO_{2-x} target. After sputtering, the samples were annealed at 350°C for 1 h in air. Subsequently, a mesoporous TiO_2 layer composed of 30-nm-sized particles was deposited by spin coating for 20 s using a commercial TiO_2 paste (Dyesol 30 NR-D) diluted in ethanol (2:7, weight ratio). After drying at 120°C for 5 min, the TiO_2 films were gradually heated to 500°C , annealed at this temperature for 30 min and cooled to room temperature. Prior to their use, the films were again dried at 500°C for 30 min.

PbI_2 was dissolved in N,N-dimethylformamide at a concentration of 462 mg/ml

(~1M) under stirring at 70 °C. The solution was kept at 70 °C during the whole procedure. The mesoporous TiO₂ films were then infiltrated with PbI₂ by spin coating for 40 s. Samples were dried at 40 °C for 5 min and 110 °C for 5 min, respectively. After cooling to room temperature, a solution of CH₃NH₃I in isopropanol (8 mg/ml) was dropped on the films and hold for 40 s, making the sample's color turn from yellow to black, then deposited by was spin coating for 20 s, and dried at 110 °C for 5 min.

The HTM was then deposited by spin coating for 30 s. The spin coating formulation was prepared by mixing solution of 72.3 mg Spiro-MeOTAD in 1 ml chlorobenzene, 28.8 μ l 4-tert-butylpyridine, and 17.5 μ l solution of TFSI in acetonitrile (520 mg/ml).

Finally, 80 nm of gold was thermally evaporated on top of the device to form the back contact. The unit devices have an active size as defined by the size of both the gold and FTO electrodes. The device fabrication was carried out under ambient conditions and a humidity of 25%.

To study the evolution characteristics of perovskite solar cells, we kept samples in two different conditions. A set of samples were stored in normal atmosphere, while the other set of samples was stored in the vacuum condition established by a turbo pump. The vacuum level is about 10⁻⁶ Torr.

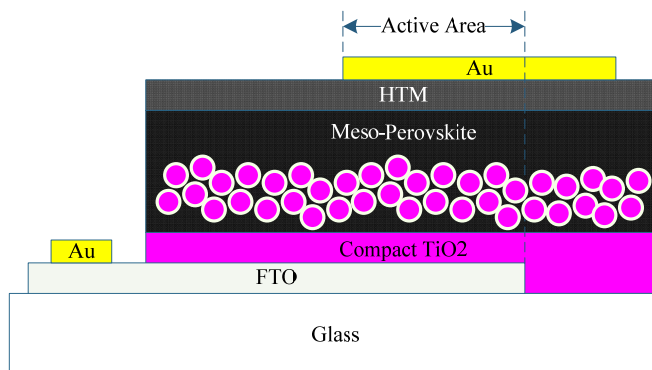


Figure 1. Cross-sectional structure of the device.

2.3 Characterizations

X-ray diffraction (XRD) analysis was performed to investigate the phases of the samples, using a SIEMENS D500. For the Scanning Electron Microscopy (SEM) measurements a FEI Helios Nanolab 660 dual beam was used. J-V characteristic curves were recorded under ambient conditions by applying an external potential bias to the cell while recording the generated photocurrent with an Agilent 4155B source meter. Illumination of the samples was made with a standard ScienceTech SS150 Solar Simulator. The ScienceTech SS150 solar simulator was calibrated to the AM1.5G standard with a standard light meter.

3. Results and discussion

3.1 SEM

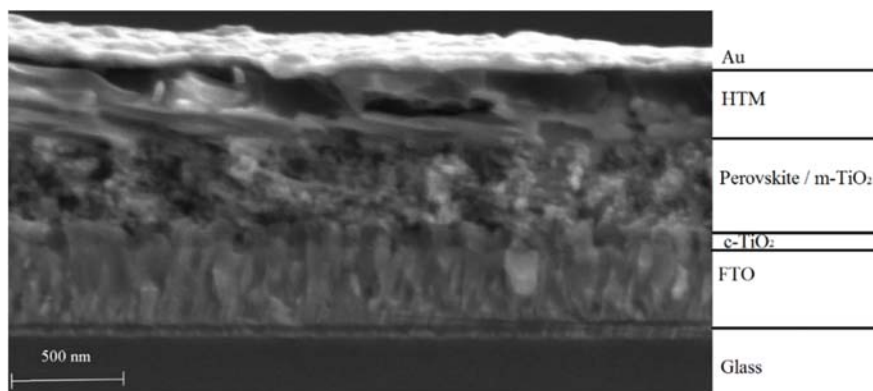


Figure 2. Cross-sectional SEM image of the perovskite film synthesized by the conventional two-step deposition method.

The device structure is shown in Fig. 1 (starting from the bottom: glass/FTO substrate, compact TiO₂ layer, mesoporous TiO₂ layer infiltrated with the perovskite nanocrystals, and hole transport layer, with a final Au electrode deposited via thermal evaporation). The results reveal that the perovskite film is uniform despite some pinholes. The cross-section SEM image shown in Fig. 2 shows a well-defined layer-by-layer structure with obvious interfaces. The thicknesses of the compact TiO₂, perovskite, and HTM layers are 60, 310, and 220 nm, respectively. As reported before, the thickness of perovskite layer around 310 nm is optimal and sufficient to serve as the light-absorbing layer^{17, 18}.

3.2 XRD

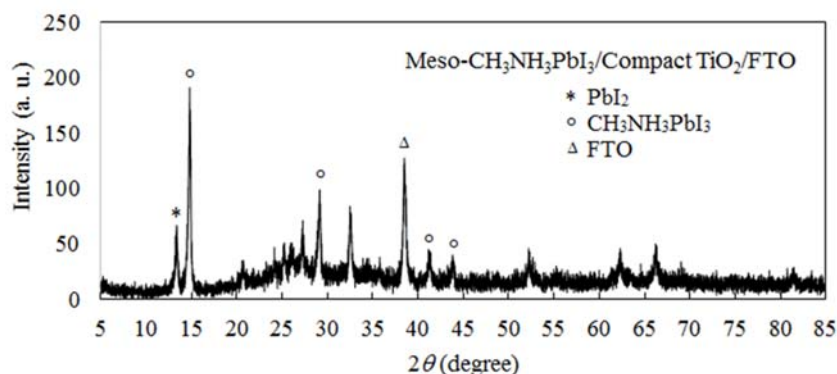


Figure 3. XRD spectra of Meso-CH₃NH₃PbI₃/Compact TiO₂/FTO.

The XRD pattern of Meso-CH₃NH₃PbI₃/Compact TiO₂/FTO is shown in Fig. 3. The main diffraction peaks centred at 14.2° (110) and 28.5° (220) can be assigned to the CH₃NH₃PbI₃ phase, showing a crystal structure, and the peak centred at 12.7° is attributed to PbI₂¹⁰. We observe unreacted PbI₂ remained, which means the conversion of PbI₂ to perovskite on exposure to the MAI solution is incomplete. This incomplete

conversion will affect the final performance of the perovskite solar cells, and more research should be carried out to solve it.

3.3 Evolution of the perovskite solar cells

The evolution characteristics measurements reveal that the vacuum plays an important role in the stability of the perovskite solar cells compared the samples in air. In order to compare the evolution characteristics intuitively, we normalized all the values with setting the initial value as 1, as shown in Fig. 4.

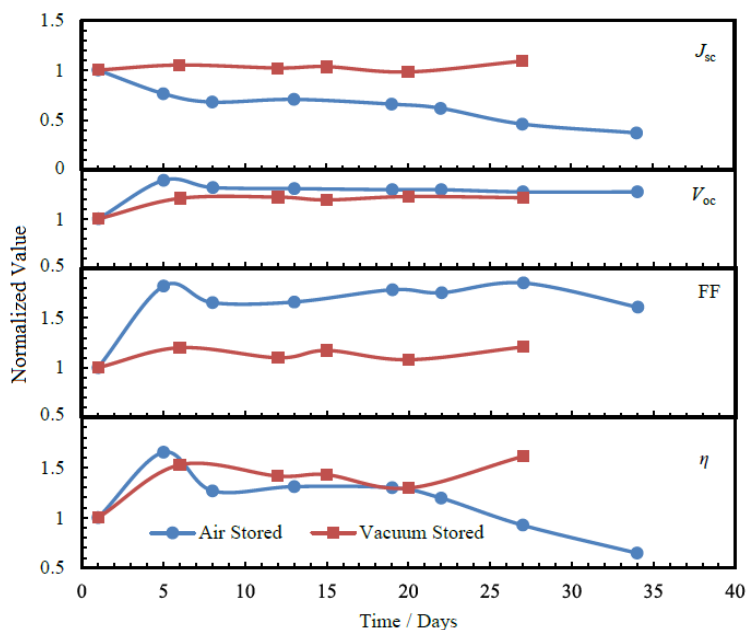


Figure 4. Evolution of perovskite solar cells stored in air and vacuum, respectively. J_{sc} , V_{oc} , FF, and PCE are normalized.

J_{sc}

Contrary to the air-stored samples without encapsulation, the J_{sc} of vacuum-stored samples is remarkably stable. The J_{sc} of air-stored samples continues to decrease for the 34 days showing significant reduction, while the J_{sc} of vacuum-stored samples remains the same level. As reported by Kim *et al*¹⁹, the J_{sc} decreases to 85% of the initial value for 8 days, where the devices are stored in air at room temperature. For our samples stored in air, there is only 68% of the initial value after 8 days for the J_{sc} , which shows a faster attenuation characteristics. For the vacuum-stored samples, the keeps constant and even rise a little after 25 days.

As well known, the light absorbing layer is the most important factor affecting the photocurrent for the solar cells. Many studies have demonstrated that the MAPbI₃ perovskite material is easily decomposed into MAI and PbI₂ in the humid environment due to the hygroscopic amine component in the compound²⁰⁻²². This reveals that the degradation of J_{sc} was mainly due to the decomposition of MAPbI₃, as a result of

reaction with H₂O²³.

V_{oc}

For both of the samples in air and vacuum, the V_{oc} shows a jump during the first 5 days. The V_{oc} of vacuum-stored samples attains a plateau after the jump, while the V_{oc} of air-stored samples shows a slightly drop and then becomes stable. The V_{oc} of air-stored samples has a larger rise compared with that of the vacuum-stored samples, and keeps a slightly larger stable value finally. Baohua Wang et al. also found the substantial increase of V_{oc} with storing the perovskite film for up to 30 days in ambient condition¹².

After the initial non-stable period, both of the V_{oc} of vacuum-stored and air-stored samples stabilized and became convergence, and the difference of V_{oc} is very small after 20 days. This performance shows that air or vacuum environment has little effect on the evolution of the V_{oc} , which means the V_{oc} of perovskite solar cells is stable under different conditions.

FF

The FF depends on the series resistance (R_s), which originates particularly from the charge transport properties of the HTM^{24,25}. As reported by Wang et al. oxidized Spiro-MeOTAD can improve the conductivity of the HTM layer, and enhance the charge transfer at the Au/Spiro-MeOTAD interface²⁶. These observations indicate that the oxidation is advantageous for improving the FF, and exposure to oxygen is a critical factor influencing solar cell performance²⁴.

During our experiments, FF of air-stored samples increased a lot for the first six days, which can be attributed to the conductivity improvement of the HTM layer caused by the oxidation of Spiro-MeOTAD. It needs 6 days to reach the maximum because the oxidization of spiro-MeOTAD is a dynamic and time-consuming process²⁴. Some oxygen exists in the vacuum chamber inevitably, so there is also a rise for the FF of vacuum-stored samples. Due to the lack of oxygen, the improvement of FF is much smaller compared to that in air.

PCE

PCE is determined by J_{sc} , V_{oc} , and FF with the equation $PCE = J_{sc} \cdot V_{oc} \cdot FF$. For the first 5 days, the PCE of air-stored samples has a great jump, which is similar as the performances reported by Liu *et al*¹¹. They observed a significant increase of PCE for the first 2 days, and the maximum value was maintained for 3 days at least before it begin to decline. This dramatic increase of PCE benefit from the improvement of FF. The PCE of vacuum-stored samples has the same trend at this rising stage, but it is slower and more modest with respect to the air-stored samples.

After the initial rising stage, PCE of air-stored samples begin to fall and continues, while the PCE of vacuum-stored samples maintain a certain level substantially. This phenomenon is due to the decomposition of perovskite material when contacts moisture

in air.

Actually, we find all the performance parameters of vacuum-stored samples change more slowly than that of the air-stored samples. With more oxygen and moisture in air, on the one hand it is good for the oxidation of HTM, which reduces the R_s and improves the FF, but on the other hand the moisture in air is deadly for the perovskite material which lead to the decrease of J_{sc} .

3.4 Best performance of the perovskite solar cells

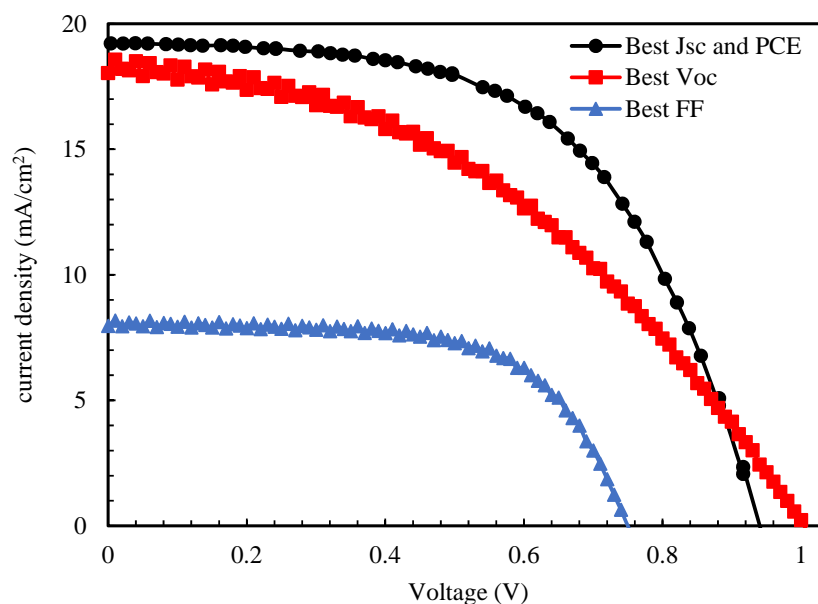


Figure 5. J - V curves corresponding to the best J_{sc} , PCE, V_{oc} , and FF.

Table 1. Solar cells performance parameters corresponding to the best values, respectively.

	PCE (%)	J_{sc} (mA/cm ²)	V_{oc} (V)	FF
Best J_{sc}	10.2	19.2	0.94	0.57
Best V_{oc}	7.78	18.0	1.0	0.43
Best FF	3.89	8.0	0.75	0.65
Best PCE	10.2	19.2	0.94	0.57

J - V curves and the performance parameters corresponding to the best J_{sc} , V_{oc} , FF, and PCE after different days are shown in Fig.5 and Table 1, respectively. Our best J_{sc} is 19.2 mA/cm², the best V_{oc} is 1.0 V, the best FF is 0.65, and the best PCE is 10.2%. It can be found that the best value for each parameter appears at the different stages during the evolution period of the perovskite solar cells. This is due to moisture and oxygen in air affecting the performance of the individual parameters in the opposite directions. FF needs a certain time to achieve the best value caused by the oxidization of HTM, while J_{sc} continues to decline caused by the decomposition of perovskite when it interacts

with moisture.

4. Conclusions

In conclusion, we compared the evolution characteristics of perovskite solar cells in air and vacuum environments. Each performance parameters of air-stored samples shows a drastic changes compared with that of the vacuum-stored samples. Moisture and oxygen in air are found to influence the perovskite solar cells performances together. HTM material needs to be oxidized to improve the conductivity, which will reduce the R_s and in favor of FF. Perovskite is very easy to decompose once contact with moisture in air, leading to the decrease of J_{sc} . Therefore, the present findings will be a reference in the storage and evolutions of perovskite solar cells, and we believe the performance can be further improved with precise control of moisture and oxygen, which will be our next research topic.

Acknowledgments

This work was substantially supported by the National Natural Science Foundation of China under Grant No. 61405025, and Central Universities Fundamental Research under Grant No. ZYGX2014J045, and the Internal fund of University of Kentucky.

References:

1. Liu, M.; Johnston, M. B.; Snaith, H. J. *Nature* **2013**, 501, (7467), 395-398.
2. Kim, H.-S.; Lee, J.-W.; Yantara, N.; Boix, P. P.; Kulkarni, S. A.; Mhaisalkar, S.; Grätzel, M.; Park, N.-G. *Nano letters* **2013**, 13, (6), 2412-2417.
3. Nie, W.; Tsai, H.; Asadpour, R.; Blancon, J.-C.; Neukirch, A. J.; Gupta, G.; Crochet, J. J.; Chhowalla, M.; Tretiak, S.; Alam, M. A. *Science* **2015**, 347, (6221), 522-525.
4. Im, J.-H.; Luo, J.; Franckevicius, M.; Pellet, N.; Gao, P.; Moehl, T.; Zakeeruddin, S. M.; Nazeeruddin, M. K.; Grätzel, M.; Park, N.-G. *Nano letters* **2015**, 15, (3), 2120-2126.
5. Grätzel, M. *Nature materials* **2014**, 13, (9), 838-842.
6. Tress, W.; Marinova, N.; Moehl, T.; Zakeeruddin, S.; Nazeeruddin, M. K.; Grätzel, M. *Energy & Environmental Science* **2015**, 8, (3), 995-1004.
7. Kojima, A.; Teshima, K.; Shirai, Y.; Miyasaka, T. *Journal of the American Chemical Society* **2009**, 131, (17), 6050-6051.
8. Yang, W. S.; Noh, J. H.; Jeon, N. J.; Kim, Y. C.; Ryu, S.; Seo, J.; Seok, S. I. *Science* **2015**, aaa9272.
9. Bi, D.; Tress, W.; Dar, M. I.; Gao, P.; Luo, J.; Renevier, C.; Schenk, K.; Abate, A.; Giordano, F.; Baena, J.-P. C. *Science Advances* **2016**, 2, (1), e1501170.
10. Yin, J.; Cao, J.; He, X.; Yuan, S.; Sun, S.; Li, J.; Zheng, N.; Lin, L. *Journal of Materials Chemistry A* **2015**, 3, (32), 16860-16866.
11. Liu, J.; Wu, Y.; Qin, C.; Yang, X.; Yasuda, T.; Islam, A.; Zhang, K.; Peng, W.; Chen, W.; Han, L. *Energy & Environmental Science* **2014**, 7, (9), 2963-2967.
12. Wang, B.; Chen, T. *Advanced Science* **2015**.
13. Noh, J. H.; Im, S. H.; Heo, J. H.; Mandal, T. N.; Seok, S. I. *Nano letters* **2013**, 13, (4), 1764-1769.
14. Bi, D.; Moon, S.-J.; Häggman, L.; Boschloo, G.; Yang, L.; Johansson, E. M.; Nazeeruddin, M. K.; Grätzel, M.; Hagfeldt, A. *Rsc Advances* **2013**, 3, (41), 18762-18766.
15. Gao, X.; Li, J.; Baker, J.; Hou, Y.; Guan, D.; Chen, J.; Yuan, C. *Chemical Communications* **2014**, 50,

(48), 6368-6371.

16. Chen, Q.; Zhou, H.; Hong, Z.; Luo, S.; Duan, H.-S.; Wang, H.-H.; Liu, Y.; Li, G.; Yang, Y. *Journal of the American Chemical Society* **2013**, 136, (2), 622-625.
17. Burschka, J.; Pellet, N.; Moon, S.-J.; Humphry-Baker, R.; Gao, P.; Nazeeruddin, M. K.; Grätzel, M. *Nature* **2013**, 499, (7458), 316-319.
18. You, J.; Hong, Z.; Yang, Y. M.; Chen, Q.; Cai, M.; Song, T.-B.; Chen, C.-C.; Lu, S.; Liu, Y.; Zhou, H. **2014**.
19. Kim, H.-S.; Lee, C.-R.; Im, J.-H.; Lee, K.-B.; Moehl, T.; Marchioro, A.; Moon, S.-J.; Humphry-Baker, R.; Yum, J.-H.; Moser, J. E. *Scientific reports* **2012**, 2.
20. Niu, G.; Li, W.; Meng, F.; Wang, L.; Dong, H.; Qiu, Y. *Journal of Materials Chemistry A* **2014**, 2, (3), 705-710.
21. Yang, J.; Siempelkamp, B. D.; Liu, D.; Kelly, T. L. *ACS nano* **2015**, 9, (2), 1955-1963.
22. Shahbazi, M.; Wang, H. *Solar Energy* **2016**, 123, 74-87.
23. Han, Y.; Meyer, S.; Dkhissi, Y.; Weber, K.; Pringle, J. M.; Bach, U.; Spiccia, L.; Cheng, Y.-B. *Journal of Materials Chemistry A* **2015**, 3, (15), 8139-8147.
24. Sheikh, A. D.; Bera, A.; Haque, M. A.; Rakhi, R. B.; Del Gobbo, S.; Alshareef, H. N.; Wu, T. *Solar Energy Materials and Solar Cells* **2015**, 137, 6-14.
25. Abate, A.; Leijtens, T.; Pathak, S.; Teuscher, J.; Avolio, R.; Errico, M. E.; Kirkpatrick, J.; Ball, J. M.; Docampo, P.; McPherson, I. *Physical Chemistry Chemical Physics* **2013**, 15, (7), 2572-2579.
26. Wang, S.; Yuan, W.; Meng, Y. S. *ACS applied materials & interfaces* **2015**, 7, (44), 24791-24798.